# **Sweet 16 Chemistry Of Gases Tournament Answers**

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# **Sweet 16 Chemistry Of Gases**

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An ideal gas adheres exactly to the kinetic theory of gases. 16.3: Pressure: The Result of Constant Molecular Collisions Pressure is a force exerted over an area. Pressure has several common units that can be converted. 16.4: Boyle's Law: Pressure and Volume Boyle's law relates a gas's pressure and volume at constant temperature and amount.

## 16: Gases - Chemistry LibreTexts

Sweet 16 Chemistry of Gases Tournament? First round: Predict the Gaseous product. Second round: Predict the one that dissolves in water to give an acidic solution. Semis: The gas that is more...

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Read PDF Sweet 16 Chemistry Gases Tournament Answers been provided to help identify the contestants in the Sweet 16 Chemistry of Gases tournament. • First round: Predict the name and formula of the gaseous product obtained in a chemical reaction involving the reactants listed in the first round. The gas "escapes" to the second round!

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First round ------ Predict the gaseous product. 1.) Na2CO3 vs. HCl 2.) Zn vs. HCl 3.) H2O2 vs. Yeast (catalyst) 4.) Na2SO3 vs. HCl 5.) Cu vs. HNO3 (concentrated) 6.) CaC2 vs. H2O 7.) NaOCl vs. HCl 8.) NH4NO2 vs. Heat 2nd round- The winner dissolves in H2O to give an acidic solution. 3rd round (semis) - The gas that is more dense (g/L) wins. Finals - The winner is a greenhouse gas produced in ...

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With spring just around the corner, your students' attention will soon be turning to spring break, sunshine, and the NCAA basketball tournament. This activity combines the popularity of the March Madness basketball pool with an overview of the periodic table, including the concepts of atomic number, atomic mass, chemical symbols, and physical properties of elements.

#### **Sweet 16 Periodic Table Tournament - Flinn**

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